



# High Energy Density Li-ion Cells for EV's Based on Novel, High Voltage Cathode Material Systems

Keith D. Kepler  
Farasis Energy, Inc.  
6-18-2014

*This presentation does not contain any  
proprietary, confidential or  
otherwise restricted information*

*Project ID: # ES213*

## Timeline

- Start Date: October, 2013
- End Date: October, 2015
- Percent Complete – 20%

## Budget

- Total Project Funding: \$3,480,000
  - DOE Share: \$2,160,000
  - FFRDC: \$600,000
  - Contractor Share: \$720,000
- 2013 Funding: ~\$300,000
- Funding for 2014: ~\$1,400,000

## Barriers

- Insufficient energy density of Li-ion battery systems for PHEV and EV applications.
- Insufficient cycle and calendar life of Li-ion battery systems.
- Accelerated energy loss at elevated voltages for Li-ion technology.

## Partners

- Argonne National Laboratory:
  - *Advanced Cathode Materials Development*
- Lawrence Berkeley National Laboratory:
  - *Advanced Cathode Materials Development*
- Dupont:
  - *High Voltage Electrolyte, Separator Development*
- Nanosys/OneD Material, LLC:
  - *High Capacity Anode Materials Development*



- New cathode and anode electrode materials and Li-ion cell components are required to enable major advances in the energy density of battery systems for transportation technologies.
- The layered and layered-layered “NMC” class of cathode materials paired against a silicon based anode offer the greatest potential to meet the PHEV and EV performance goals.
- Utilization of the inherent capacity in these systems can be greatly increased if higher voltage operation can be enabled.
- There are multiple interacting failure mechanisms at the materials and cell level that are barriers to achieving the system level battery performance goals.
- A focus on cell level development utilizing advanced materials and components is critical to achieving major breakthroughs in battery performance.

# Relevance - Project Objectives

## Project Goal:

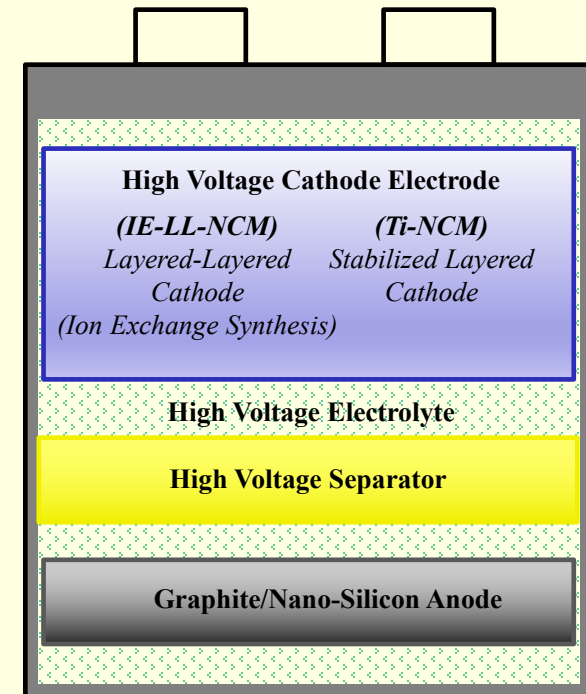
The goal of this project is to develop and demonstrate new high energy, high voltage capable Li-ion materials and cell components to enable high energy, high power Li-ion cells that have the potential to meet the performance goals of PHEV40 and EV light-duty vehicles.

## Performance Objective:

The objective is to demonstrate a PHEV40 cell with an energy density of 250 Wh/kg and an EV light duty cell with an energy density 350 Wh/kg that can meet the cycle life goals for those applications.

### Cell Level Goals:

Energy Storage Requirements			
Characteristics	Unit	PHEV40	EV
Specific Discharge Pulse Power	W/kg	800	800
Discharge Pulse Power Density	W/l	1600	1200
Specific Regen Pulse Power	W/kg	430	400
Regen Pulse Power Density	W/l	860	600
Recharge Rate		C/3	C/3
Specific Energy	Wh/kg	200	400
Energy Density	Wh/l	400	600
Calendar Life	Year	10+	10
Cycle Life (at 30°C with C/3 charge and discharge rates)	Cycles	5,000	1,000
Operating Temperature Range	°C	-30 to +52	-30 to +65



**Project Technical Targets** **Year 1 (Gen 1):**  
*Cell Level 230 Wh/kg, 1000 cycles (PHEV)*

**Year 2 (Final Deliverable Cells):**  
*Cell Level 250 Wh/kg, 5000 cycles (PHEV), Cell Level 350 Wh/kg, 1000 cycles (EV)*



# First Year Technical Milestones

- Milestones leading to cell performance data from “Gen 1” cell build representing first try at integrating improved materials, electrolytes and cell designs.

## Cell Build Progression:

Year 2



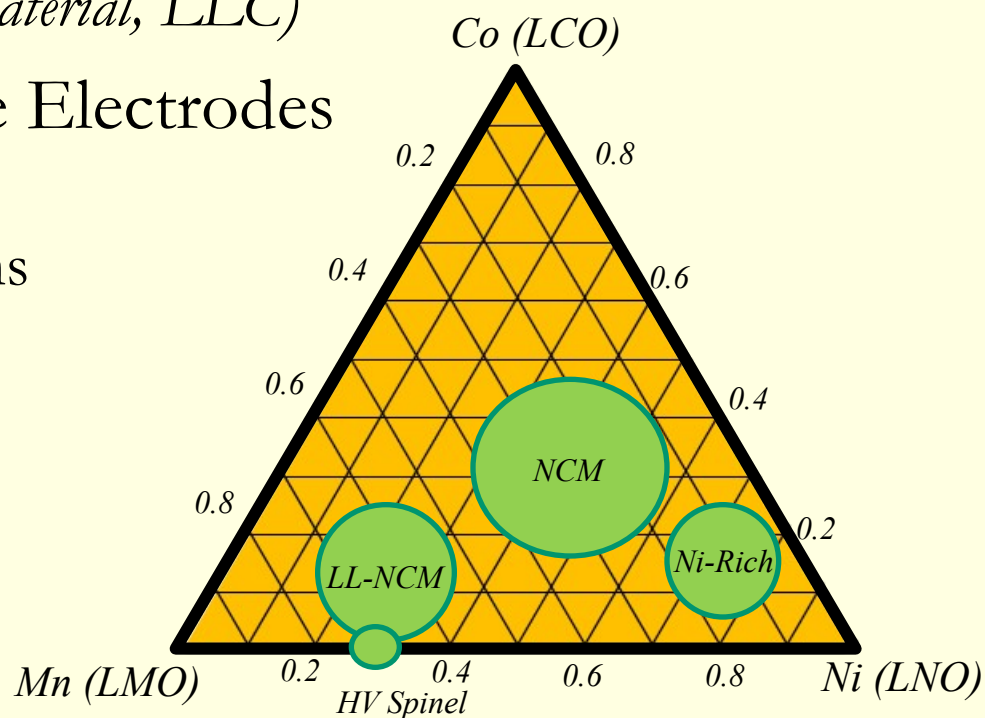
## First Year Milestones and Status

Month	Milestone	Type	Description
Month 5	Completion of Baseline Cell Deliverables	Technical	High Voltage LL-NCM/Graphite pouch cells
Month 9	Completion of Round 1 Electrolyte Evaluation	Technical	Successful identification of candidate high voltage stable electrolyte formulations.
Month 11	Selection of GEN 1 Cathode Materials Complete	Technical	LLS-NCM, Ti-NCM, & Si Anode material selected using high voltage stable electrolytes
Month 12	Completion of GEN 1 Small Cell Build	Go/No Go	Energy density exceeding of 230 Wh/kg at a C/3 rate estimated from testing in a 200 mAh form factor with projected cycle life of > 1000 cycles

# Technical Approach

Development focused on addressing key current barriers to achieving high capacity long life Li-ion cells.

- Higher Capacity, Higher Voltage Active Materials
  - IE-LLS-NCM (*Argonne National Laboratory*)
  - Stabilized-NCM (*Lawrence Berkeley National Laboratory*)
  - Si-Graphite Composite (*OneD Material, LLC*)
- Higher Rate Capability Cathode Electrodes
  - Ion Exchange Synthesis
  - Composite Cathode Formulations
- Higher Voltage Operation
  - Cathode Surface Stabilization
  - Stable Electrolytes (*Dupont*)
  - Stable Separator (*Dupont*)



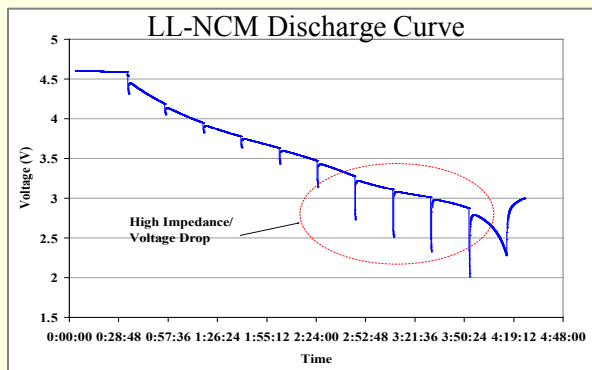
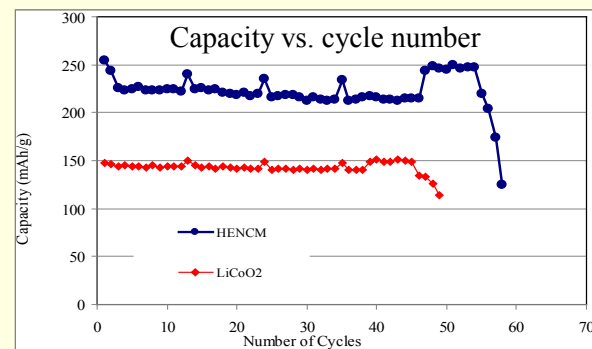
# Technical Approach Layered-Layered NCM

## Advantages:

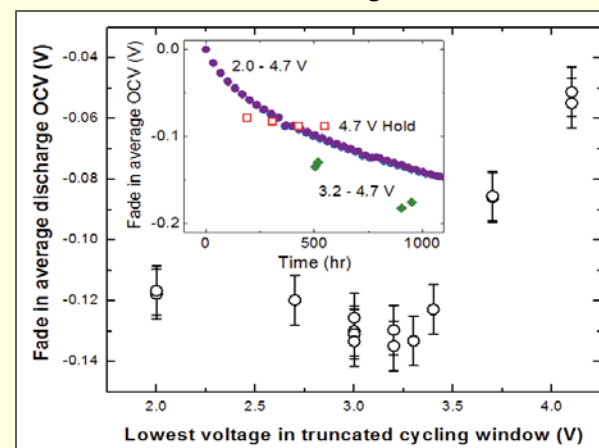
- High specific capacity – 230-250 mAh/g.
- Greater stability at high voltages.

## Barriers:

- High impedance.
- State of charge dependent impedance and impedance growth.
- Voltage fade mechanism.
- Accelerated capacity loss if not stabilized.
- Low utilization in full cells.
- Low tap density.
- Wide voltage window.



OCV drop during cycling LL-NCM within different voltage windows



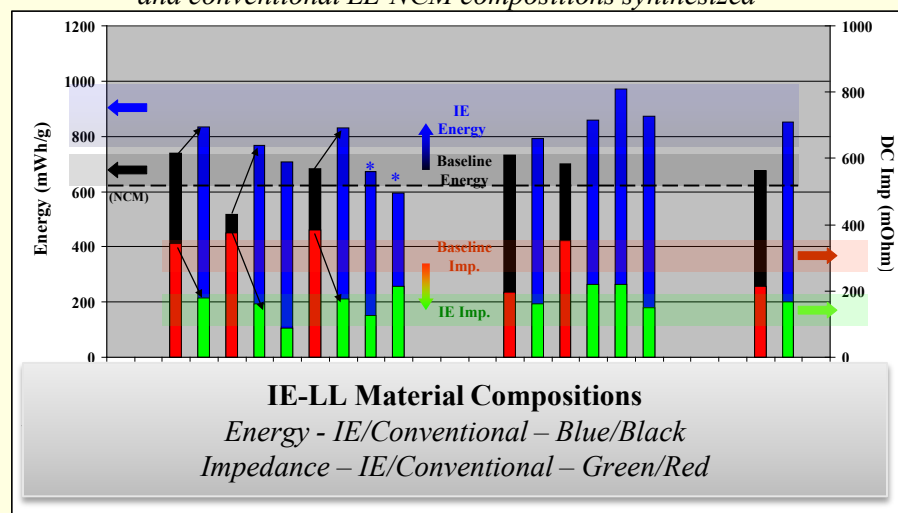


# Technical Approach

## Layered-Layered NCM

- Development strategy based on initial work done by Dr. Chris Johnson at Argonne National Laboratory and continued at Farasis Energy.
- Ion-Exchange Synthesis Approach
  - Na based LL-NCM material is used as a precursor to form Lithium LL-NCM through an ion-exchange process with Lithium (IE-LL-NCM)
  - Composition and synthetic conditions can be tuned to produce a high voltage spinel component to the LL materials → Layered-Layered-Spinel NCM (LLS-NCM)
  - Initial work indicates synthetic approach leads to materials with lower impedance and greater utilization.
- Potential for New Structural and Performance Characteristics
  - Potential to avoid O3 stacking and transition metal movement during cycling.
  - Route to creation of materials with larger interlayer spacings.
  - Route to introduce disorder into materials.
  - Route to materials with different surface morphology, stacking faults.

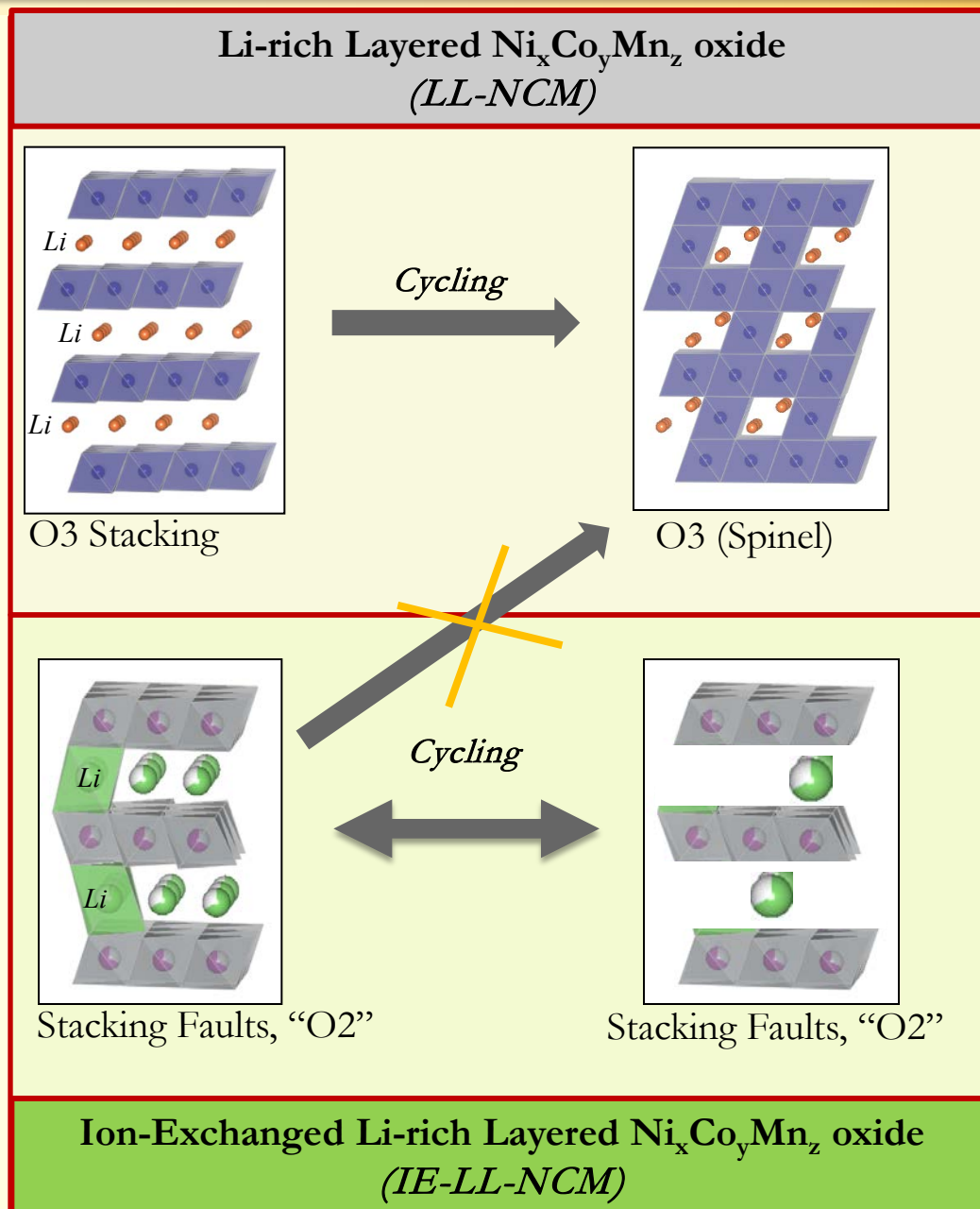
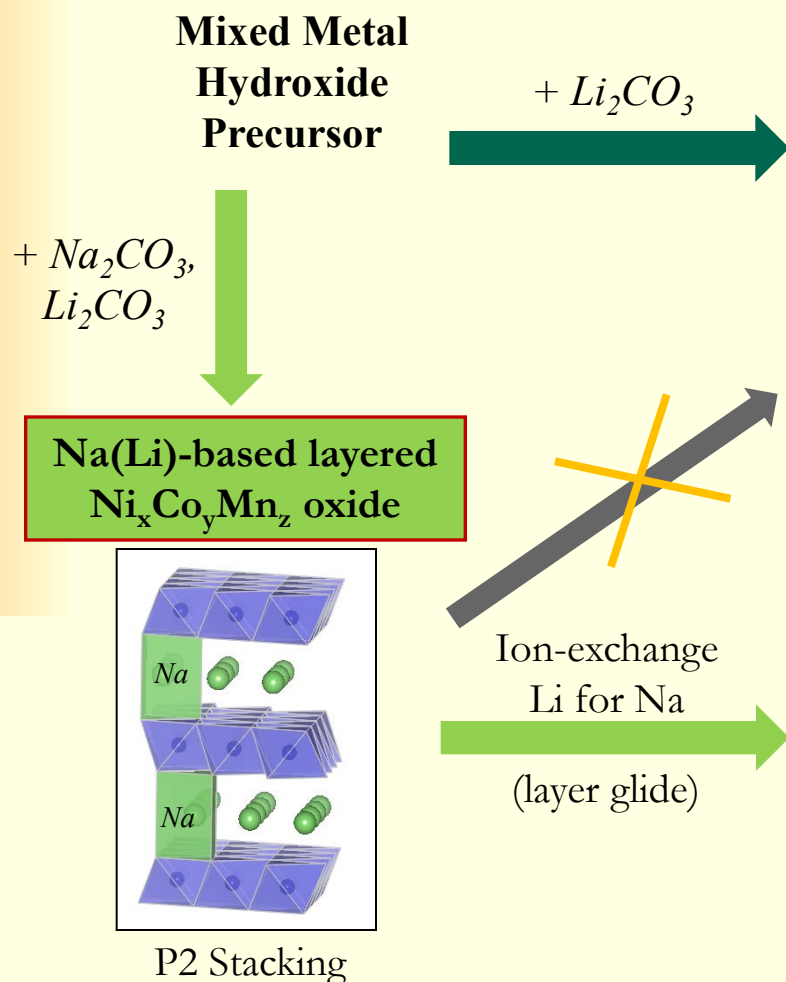
Comparison of energy and impedance measured for a number of IE and conventional LL-NCM compositions synthesized





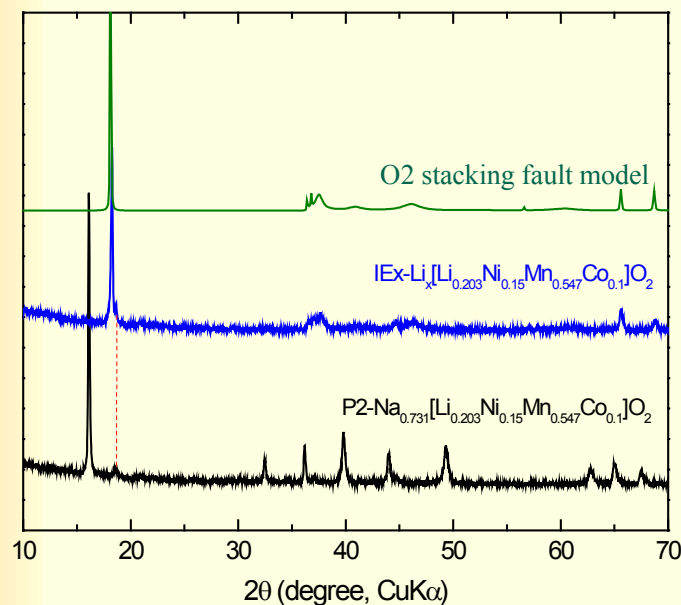
# Ion-Exchange Synthesis of Layered-Layered NCM

*Possible impact of ion-exchange route on structure of high energy materials.*

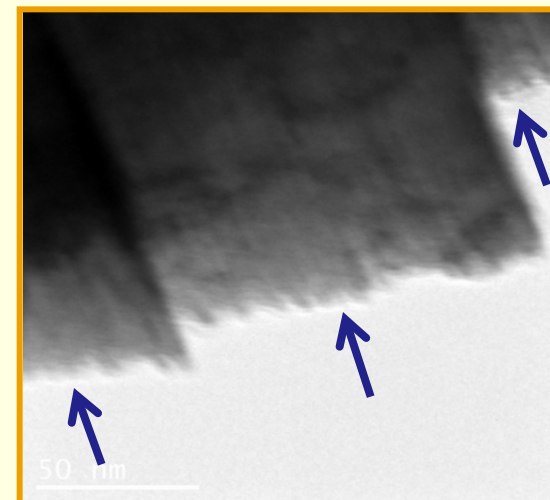
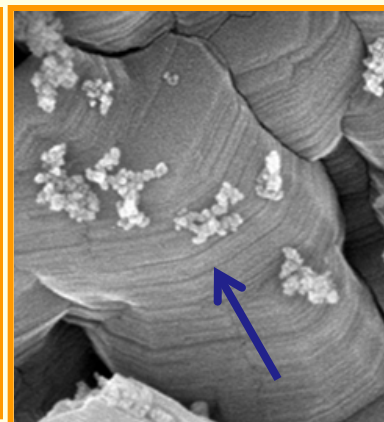
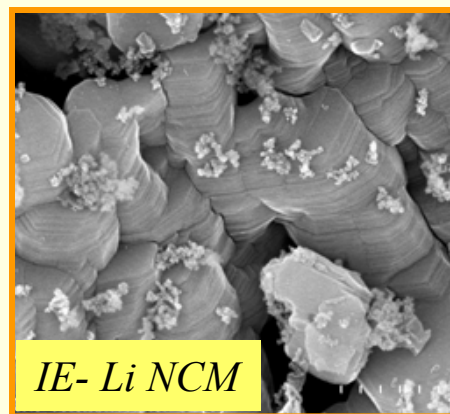
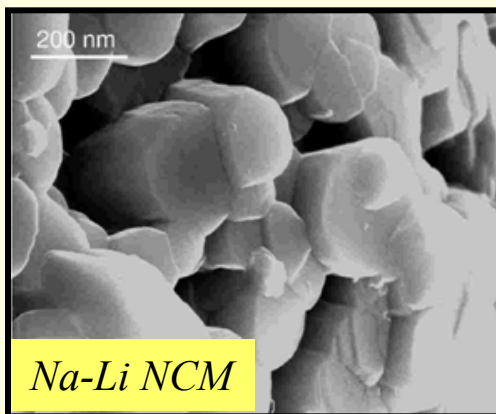
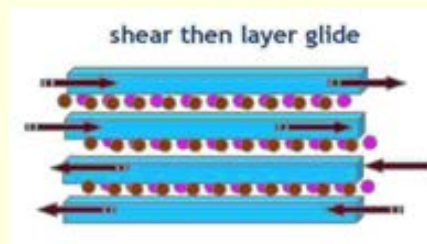


# Ion-Exchange Synthesis of Layered-Layered NCM

- X-ray diffraction indicates good layering order but significant disorder in other crystallographic directions suggesting presence of stacking faults.



- Faults in shear order of crystal lattice during ion exchange
- Still strongly layered
- Local c-axis disorder
- Structural modeling indicates presence of extensive stacking faults with O2 layering characteristics.



# Technical Approach

## Layered NCM Materials

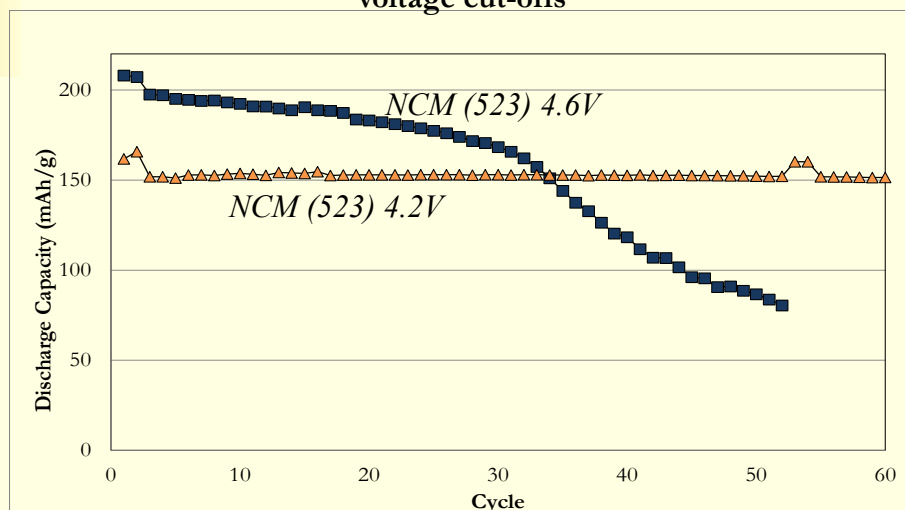
### Advantages:

- Good rate capability
- High tap density
- Good stability at moderate voltages
- Reasonable average voltage

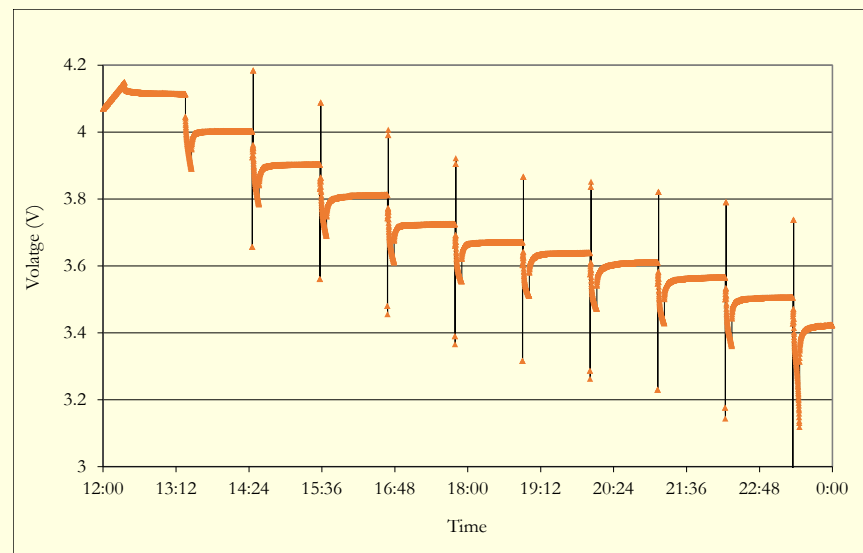
### Barriers:

- Stability at high voltages.

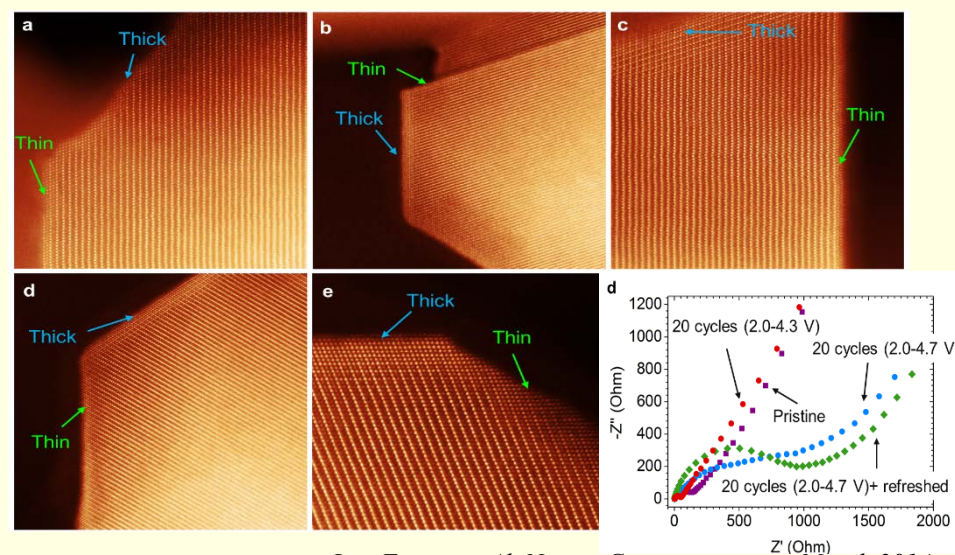
Relative stability of NCM (523) cathode to different upper voltage cut-offs



NCM/Graphite Cell HPPC test



Rock-salt surface reconstruction occurs upon electrolyte exposure alone, but is more severe when electrodes are cycled to 4.7V



# Technical Approach

## Layered NCM Materials

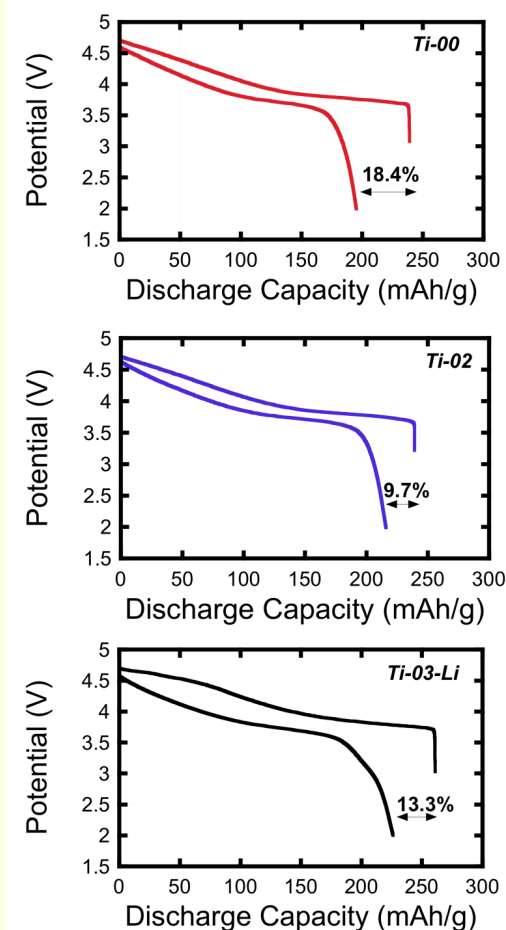
### Surface Stabilization:

- Coatings/surface treatments.
- Decrease active material surface reactivity to electrolyte.

### Doping:

- Bulk addition of elemental dopants to NCM composition.
- Stabilize layered structure in highly charged state.
- Aliovalent substitution to limit oxygen loss/surface reconstruction.

High Voltage Formation  
Curves of Ti-Doped NCM(424)



Kam, Kinson C., et. Al, J. Mater. Chem, 2011, 21 9991.

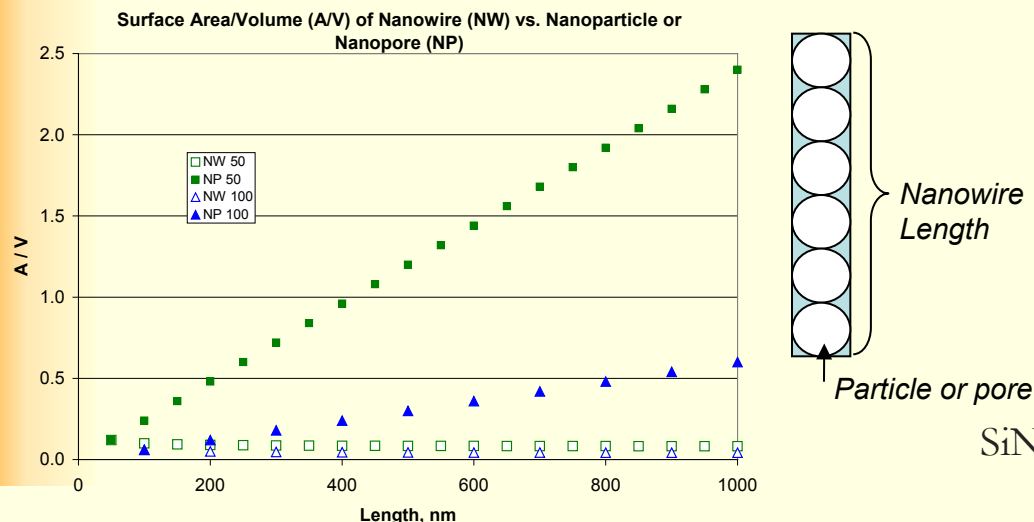


# Technical Approach

## Nano-Silicon Anode Materials

### Nanosys SiNANOde Approach vs. Hollow/Porous Approach

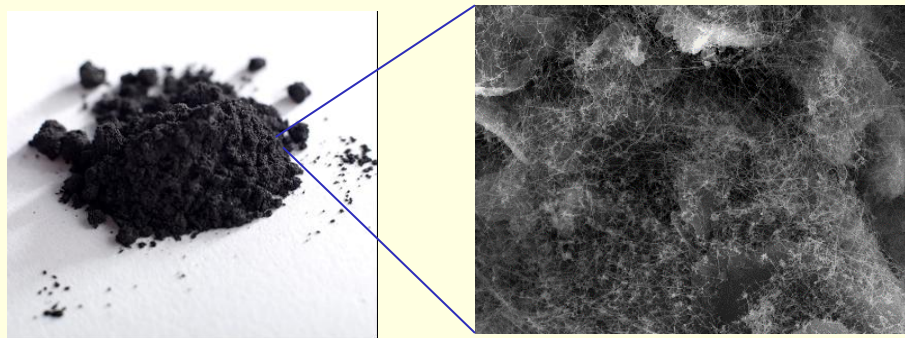
SiNANOde	Hollow/Porous Si
Low A/V & Intact NW after cycling	High A/V; defects
Pack density similar to graphite	Pack density lower than graphite
Mass-produced with a competing cost * high Si utilization	Difficult and expensive to commercialize



- A Si nanowire is equivalent to several Si particles or pores with an identical diameter.
- Si nanowire has lower surface area/volume ratio (A/V) and hence less side-reaction with electrolyte and better cycle life

SiNANOde production process: Directly grow Si nanowires on graphite powders

- Cost effective and high Si utilization
- Improves dispersion in slurry and drop in process (just replace graphite powders)
- Si-C conductivity improvement
- Si% or anode specific capacity is controllable, focusing on **500 ~ 1600 mAh/g**
- High electrode loading, as high as 1.5g/cm<sup>3</sup>
- Good cycling performance, cycled >1000 times





# Technical Approach

## High Voltage Li-ion Cell

---

- Develop high voltage capable fluorinated electrolytes with proper battery system design to enable operation up to 4.7 V:
  - Increase cell **Energy Density** by enabling higher voltages
  - Increase cell **Power Density** by maintaining/improving conductivity
  - Lower System **Costs** by enabling higher voltages, reducing number of cells needed and potentially simplifying packaging requirements
  - Good wettability will drive similar manufacturing processes
- Incorporation of separators that are inherently stable to high voltage operation.
- Improve adhesion stability of electrode laminates.
- Incorporation of low reactivity electrode laminate components.

# Technical Accomplishments: Baseline Deliverable Cells

*Milestone 1 (Month 5): Completion of Baseline Cell Deliverables*

- LL-NCM/Graphite Li-ion Pouch Cells
- Capacity  $\sim 1.6$  Ah
- Fourteen cells shipped to Idaho National Laboratory.
- Developing test plan with INL and DOE program managers.



Baseline Pouch Cell





# Technical Accomplishments: High Voltage Electrolyte Development

## *Milestone 2 (Month 6): Completion of “Gen 0” Cell Builds*

- “Gen 0” Cells for first round electrolyte development:
  - 18650 and pouch cells using conventional NCM/Graphite based chemistries designed for 4.4V and 4.6V operation.
  - Pouch cells using early stage advanced chemistries and silicon anodes.
- Ongoing work involves formulation optimization, formation protocols studies, failure mode analysis, gas generation measurements, accelerated testing.
- Have achieved significant gains in stability at high voltages relative to baseline carbonate electrolytes.

*“Gen 0” Cells*

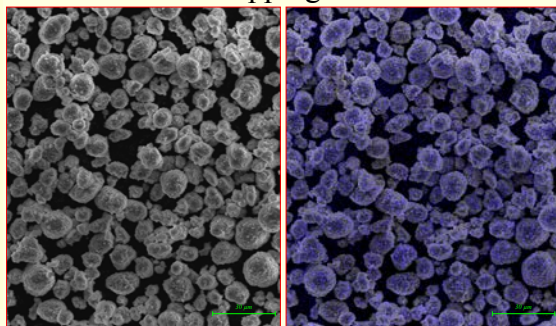


# Technical Accomplishments

## Materials Development

- Received and characterized baseline materials from each subcontractor.
- Established uniform evaluation protocol for new cathode materials.
- Stabilized NCM –
  - Performed initial experiments to evaluate feasibility of low cost synthetic routes of doped NCM compositions.
  - Successfully scaled surface stabilization process for several NCM cathode compositions (0.5kg batches).
  - Cell design and initial evaluation of stabilized NCM materials at high voltage in full pouch cells.

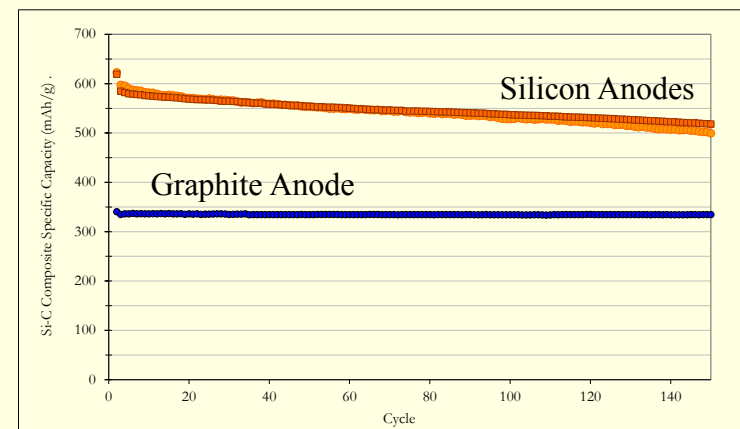
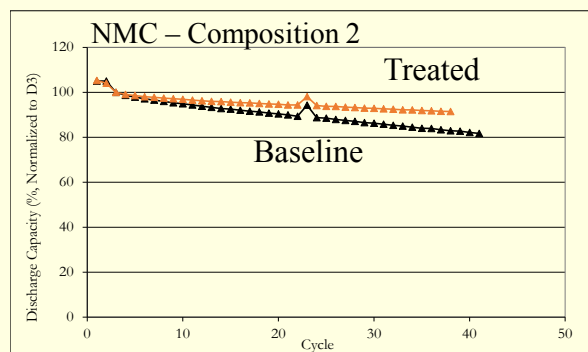
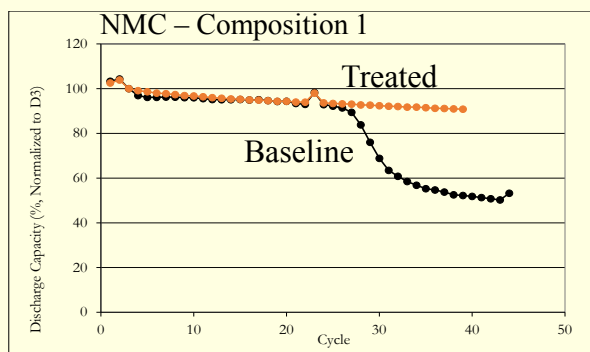
SEM/EDS elemental mapping of NCM surface treatment



### Silicon Anode Development

- Optimization of Si/Graphite electrode formulations.
- Initial full cell evaluation against baseline cell cathodes.

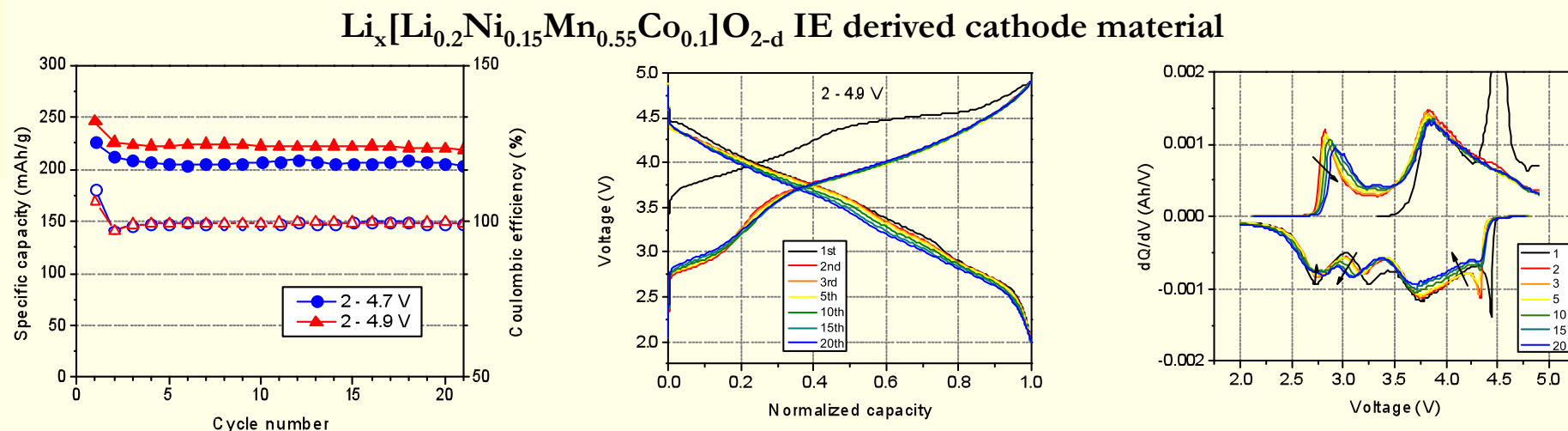
Initial pouch cell evaluation of stabilized NCM materials at 4.6V



# Technical Accomplishments

## Materials Development

- Ion-exchanged LL-NCM
  - Started evaluation of compositional effects on material performance and structure.
  - Initiated experiments for comparison of lab scale ion-exchange materials against corresponding LL-NCM materials synthesized by conventional methods.
  - Initiated work on scaling ion-exchange processes.
  - Identified and are addressing issues associated with scaling ion-exchange processes.





# Collaborations and Coordination with Other Institutions

---

## **Argonne National Laboratory (Chris Johnson)**

Federal Laboratory – Subcontractor providing materials and analytical work for project.

- Layered-Layered-(Spinel) (LL-S) NCM Cathode Material Development – Developing an ion-exchange synthetic approach to address the impedance and voltage fade barriers of high capacity LL-NCM cathode materials.

## **Lawrence Berkeley National Laboratory (Marca Doeff):**

Federal Laboratory – Subcontractor providing materials and analytical work for project.

- High Voltage Stabilized NCM Cathode Material Development – Develop and optimize doping and advanced coating methods to stabilize high capacity NCM materials to operation at high voltages.

## **Nanosys/OneD Material, LLC (Yimin Zhu):**

Industry – Subcontractor providing materials and development guidance for project.

- Nano-Silicon Graphite Composite Anode Material Development – Optimize nano-silicon graphite composites for long term cycling stability.

## **Dupont (Srijanani Bhaskar):**

Industry – Partner providing materials and analytical work for project.

- High Voltage Capable Electrolytes and Cell Components- Develop new fluorinated electrolyte systems, additives and separators with exceptional high voltage stability to advanced active materials.

# Proposed Future Work

- Complete high voltage cell component development and evaluation in Gen 0 cells.
- Characterize failure mechanisms associated with high voltage cell operation in Gen 0 chemistry.
- Develop and optimize ion-exchange LL-NCM compositions for capacity, rate capability and stability.
- Perform detailed structural and electrochemical characterization of new materials and impact of compositional and synthesis variables on material.
- Evaluate new IE-LL-NCM materials using “voltage fade” protocols.
- Develop synthetic methods for making aliovalent doped NCM materials.
- Demonstrate improved stability and performance characteristics of stabilized NCM materials at elevated voltages.
- Select and scale synthesis of best materials for Gen 1 cell build.
- Further optimization of Silicon anode electrode for Gen 1 cell build.
- Plan Gen 1 cell build and design, build and test cells.

# Summary Slide

---

- Project is relevant to the development of high energy Li-ion cells capable of meeting the PHEV40 and EV performance goals set out by DOE.
- Approach to addressing current cell level performance barriers based on strong advanced materials technical foundation.
  - Ion exchange synthetic approach to address impedance and voltage fade issues.
  - Doping and surface stabilization to improve high voltage stability.
  - Cell component development aimed at enabling long term high voltage operation.
- Strong coordination with subcontractors and partners with steps taken to allow parallel development of multiple cell components for incorporation into high performance cells.
- Future work will continue advanced cathode and anode material development and optimization leading to Gen 1 cell build at the end of Year 1.